

D 102128

(Pages : 2)

Name.....

Reg. No.....

**SECOND SEMESTER M.Sc. (CBCSS) REGULAR/SUPPLEMENTARY DEGREE  
EXAMINATION, APRIL 2024**

Chemistry

CHE 2C 08—ELECTRO CHEMISTRY, SOLID STATE CHEMISTRY AND STATISTICAL  
THERMODYNAMICS

(2019 Admission onwards)

Time : Three Hours

Maximum Weightage : 30

**Section A***Answer any eight questions.**Each question carries weightage 1.*

- Write electrode reactions in methanol fuel cell.
- Devise an electrochemical reaction in which the reaction  $\text{AgBr}_{(s)} \rightarrow \text{Ag}^+ + \text{Br}^-$  is taking place.
- Write Helmholtz model of electrical double layer. What are its drawbacks ?
- Explain the term 'electrode polarization'.
- Write Hermann-Mauguin symbol for (a)  $C_{2h}$  ; (b)  $D_{2d}$ .
- A plane cuts the  $x$ ,  $y$  and  $z$  axes at  $3a$ ,  $2b$  and  $1c$ . Write the corresponding Miller indices.
- Explain with example 'color centers' in solids.
- Explain the term 'most probable distribution'. How would you identify it ?
- Write symmetry number for (a)  $\text{CH}_4$  ; (b) Ethylene.
- State and explain equipartition principle.

(8 × 1 = 8 weightage)

**Section B***Answer any six questions.**Each question carries weightage 2.*

- Write Debye Hückel limiting law. How is it verified experimentally ?
- The EMF of the cell  $\text{Pt} \left| \text{H}_2 \right| \text{HCl} \left| \text{AgCl}_{(s)} \right| \text{Ag}$  is 0.3524 V at 25°C. The standard electrode potential of  $\text{Cl}^- \left| \text{AgCl}_{(s)} \right| \text{Ag}$  is 0.2224 V. Calculate the mean ionic activity coefficient of 0.1 molal HCl.

Turn over

13. Briefly discuss one of the theories of hydrogen over voltage.
14. Draw stereographic projection for a triclinic system. Discuss.
15. Briefly discuss working of a two stage laser.
16. Derive equations to show the relationship between partition function and (a)  $U$  ; (b)  $S$ .
17. Calculate the heat capacity of diamond at 1000 K. Its characteristic temperature is 1860 K.
18. Derive Fermi Dirac distribution law.

(6 × 2 = 12 weightage)

### Section C

*Answer any two questions.  
Each question carries weightage 5.*

19. Derive Butler-Volmer equation. Discuss.
20. Briefly discuss Bore-Einstein condensation.
21. Briefly discuss band theory of solids.
22. What are the drawbacks of Einstein's theory of heat capacity of solids ? How are they overcome by Debye ? Discuss.

(2 × 5 = 10 weightage)